

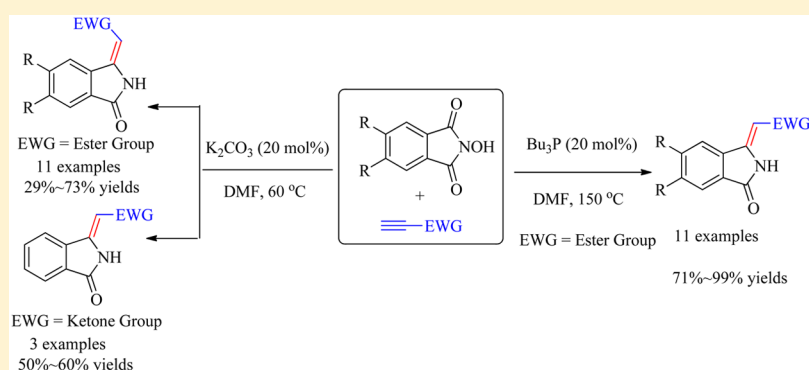
Stereoselective Synthesis of 3-Methyleneisoindolin-1-ones via Base-Catalyzed Intermolecular Reactions of Electron-Deficient Alkynes with *N*-Hydroxyphthalimides

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Supporting Information



ABSTRACT: Highly stereoselective intermolecular reactions of electron-deficient alkynes with *N*-hydroxyphthalimides for efficient construction of *N*-unprotected 3-methyleneisoindolin-1-ones have been developed through base catalytic strategies. The reaction of alkyne esters with *N*-hydroxyphthalimides catalyzed by Bu_3P in DMF at 150 °C gave the corresponding 3-methyleneisoindolin-1-ones with a (*Z*)-configuration, while the reaction of alkyne esters with *N*-hydroxyphthalimides catalyzed by K_2CO_3 in DMF at 60 °C gave the corresponding 3-methyleneisoindolin-1-ones with an (*E*)-configuration, and (*Z*)-3-methyleneisoindolin-1-ones were obtained when alkyne ketones reacted with *N*-hydroxyphthalimide.

Isoindolinones are important scaffolds in organic chemistry and medicinal chemistry due to their prevalence in numerous synthetic and naturally occurring bioactive molecules. In particular, 3-methyleneisoindolin-1-ones have been recognized as core structures in natural compounds such as enterocarpam II, the secophthalide–isoquinoline ene-lactam fumaridine,¹ magallanesine, an isoindolobenzazocine isolated from the South-American plant *Berberis darwinii*.² 3-Methyleneisoindolin-1-ones have shown a diverse range of bioactivities, such as vasorelaxant property³ and local anesthetic activity.⁴ In addition, 3-methyleneisoindolin-1-ones are also important intermediates for the synthesis of other useful alkaloids.⁵ Owing to their great importance, many methods have been developed for the preparation of 3-methyleneisoindolin-1-ones. These methods mainly involved the traditional condensation reaction of phthalimides with stabilized phosphoranes,⁶ or addition of organometallic reagents, followed by dehydration of the resulting 3-hydroxyphthalimides,⁷ the Horner condensation of 3-(diphenylphosphinoyl)isoindolin-1-ones with aldehydes,⁸ ortholithiation–anionic cyclization of *N*-acyl-2-bromobenzamides,⁹ electrophilic cyclizations of 2-alkynylbenzamides,¹⁰ and recently developed metal-catalyzed cascade reactions.¹¹ However, most of the present procedures suffer from one or more drawbacks such as

the use of expensive metal reagents or poor stereoselectivity of the products; therefore, more efficient and simple protocols to stereoselectively prepare 3-methyleneisoindolin-1-one would be highly desirable.

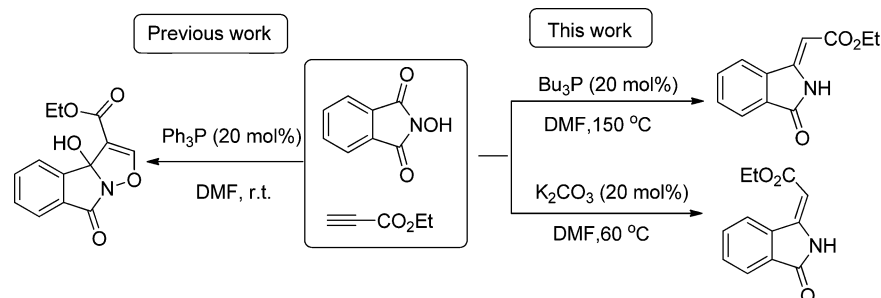
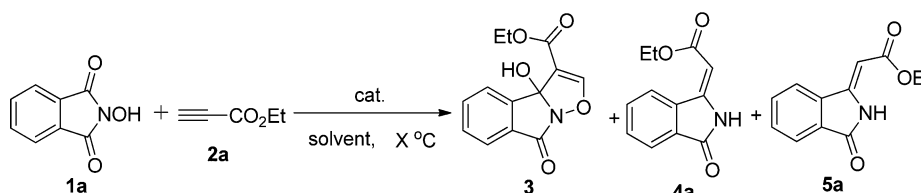
Over the past 20 years, base-catalyzed annulation has grown significantly and represents one of the most powerful synthetic methodologies for preparing various heterocyclic compounds. Nonetheless, base catalytic, direct syntheses of 3-methyleneisoindolin-1-one derivatives have seen considerably less progress. We have recently developed a convenient phosphine-catalyzed [3 + 2] annulation of electron-deficient alkynes with *N*-hydroxyphthalimide (NHPI) for the synthesis of 3-hydroxyisoxazolo[3,2-*a*]isoindol-8(3*aH*)-ones (Scheme 1).¹² In the context of ongoing projects for the diversely functionalized construction of isoindolin-1-ones, we wish to report the novel process for stereoselective synthesis of 3-methyleneisoindolin-1-ones (Scheme 1).

We started our investigation by examining the Bu_3P -catalyzed reaction of 2-hydroxyisoindoline-1,3-dione **1a** and ethyl propiolate **2a** (Table 1). First, the reaction of **1a** with **2a**

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Scheme 1. Phosphine-Catalyzed Reactions of 2-Hydroxyisoindoline-1,3-dione and Ethyl Propiolate

Table 1. Evaluation of Conditions for the Reaction of 1a with Ethyl Propiolate (2a)^a

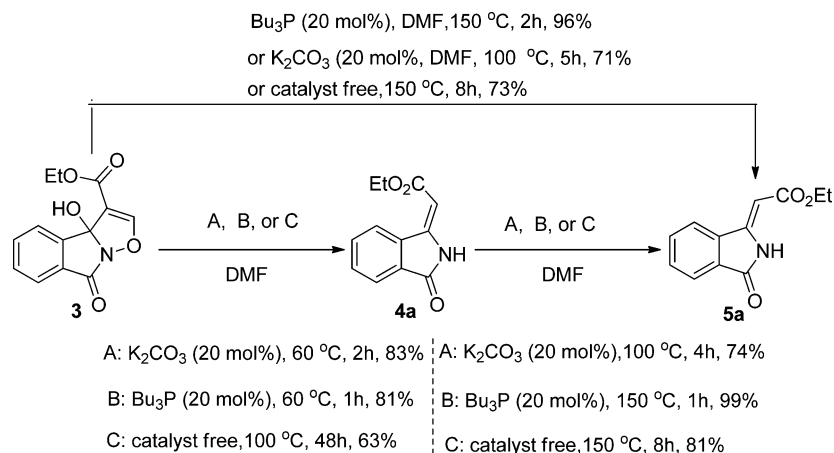
entry	catalyst (mol %)	solvent	temperature (°C)	time (h)	yield of 3 (%) ^b	yield of 4a (%) ^b	yield of 5a (%) ^b
1	Bu ₃ P (10)	DMF	r.t. ^c	48	48	42	0
2	Bu ₃ P (20)	DMF	r.t.	24	0	69	11
3	Bu ₃ P (50)	DMF	r.t.	2	0	76	6
4	Bu ₃ P (20)	DMF	0	48	40	36	14
5	Bu ₃ P (20)	DMF	-20	6	97	0	0
6	Bu ₃ P (20)	DMF	60	6	0	74	17
7	Bu ₃ P (20)	DMF	100	2	0	80	11
8	Bu ₃ P (20)	DMF	150	2	0	0	95
9	Bu ₃ P (20)	THF	reflux	48	0	0	0
10	Bu ₃ P (20)	CH ₂ Cl ₂	r.t.	48	18	28	5
11	Bu ₃ P (20)	CH ₂ Cl ₂	reflux	48	38	35	11
12	Bu ₃ P (20)	CH ₃ CN	r.t.	48	31	49	10
13	Bu ₃ P (20)	CH ₃ CN	reflux	12	0	65	13
14	Bu ₃ P (20)	toluene	r.t.	48	30	19	trace
15	Bu ₃ P (20)	toluene	reflux	12	0	51	21
17	Ph ₃ P (20)	DMF	100	18	0	40	31
18	Ph ₃ P (20)	DMF	150	16	0	0	62
19	Ph ₂ PtEt (20)	DMF	100	18	0	51	42
20	Ph ₂ PtEt (20)	DMF	150	14	0	0	92
21 ^d	Et ₃ N (20)	DMF	60	4	0	0	0
22	K ₂ CO ₃ (20)	DMF	60	4	0	73	trace
23	K ₂ CO ₃ (20)	DMF	100	6	0	0	74

^aUnless noted otherwise, reaction of *N*-hydroxyphthalimide **1a** (0.3 mmol), ethyl propiolate **2a** (0.36 mmol) was performed in 1 mL of solvent under N₂. ^bIsolated yield based on **1a**. ^cr.t. = room temperature. ^dThe Michael addition product was formed in 77% yield.

in the presence of Bu₃P (10 mol %) in DMF at room temperature for 48 h afforded the (*E*)-3-methyleneisoindolin-1-one **4a** in the yield of 42% accompanied by 3a-hydroxyisoxazolo[3,2-*a*]isoindol-8(3a*H*)-one derivative **3** in the yield of 48% (entry 1, Table 1). It was interesting to note that 3a-hydroxyisoxazolo[3,2-*a*]isoindol-8(3a*H*)-one derivative **3** was not found and the (*E*)-3-methyleneisoindolin-1-one **4a** was formed in good yield beside a little amount of (*Z*)-3-methyleneisoindolin-1-one **5a** when the loading of the catalyst Bu₃P was increased to 20 or 50 mol % (entries 2 and 3, Table 1). To improve the reaction selectivity, the temperature was then evaluated. To our delight, an excellent yield of **5a** was obtained when the reaction was carried out at 150 °C (entry 8, Table 1). The (*E*)-3-methyleneisoindolin-1-one **4a** was formed in good yield accompanied by a small quantity of its (*Z*)-isomer

5a when the reaction was disposed at 60 °C for 6 h or 100 °C for 2 h (entries 6 and 7, Table 1). However, only 3a-hydroxyisoxazolo[3,2-*a*]isoindol-8(3a*H*)-one derivative **3** was given when the reaction was performed at -20 °C (entry 4, Table 1). When shifting the solvent to THF, CH₂Cl₂, CH₃CN, or toluene, the reaction did not give the better results (entries 9–15, Table 1). Other nucleophilic phosphines were also tested, and both Ph₃P and Ph₂PtEt were proved to be useable catalysts for **5a** at 150 °C (entries 17–20, Table 1). The base containing nitrogen, such as Et₃N, was also examined in the reaction; only the Michael addition product was observed (entry 21, Table 1). It is highlighted here that only (*E*)-3-methyleneisoindolin-1-one **4a** was given in 73% yield when the reaction was performed in DMF at 60 °C for 4 h using inorganic base K₂CO₃ as catalyst, and the sole product **5a** was formed in 74%

Scheme 2. Transformation of Isoindolinone Derivatives at Different Conditions

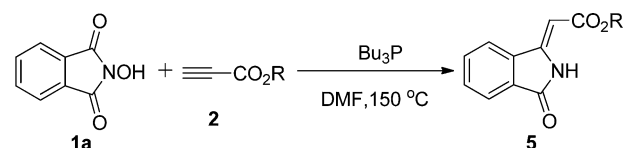


yield when the reaction was performed in DMF at 100 °C for 6 h (entries 22 and 23, Table 1). Herein, we established the optimal reaction conditions for the stereoselective synthesis of N-unprotected 3-methyleneisoindolin-1-one derivatives via the reaction conditions' control.

To understand these novel processes, 3a-hydroxyisoxazolo[3,2-*a*]isoindol-8(3*aH*)-one derivative **3** and (*E*)-3-methyleneisoindolin-1-one **4a** were treated under different conditions (Scheme 2). We found that compound **3a** could be effectively transformed to (*E*)-3-methyleneisoindolin-1-one **4a** in the presence of a catalytic amount of both K_2CO_3 and Bu_3P at 60 °C, while it was fully transformed to (*Z*)-3-methyleneisoindolin-1-one **5a** at 150 °C in the presence of Bu_3P or at 100 °C in the presence of K_2CO_3 . We also found that (*E*)-3-methyleneisoindolin-1-one **4a** could be effectively transformed to (*Z*)-3-methyleneisoindolin-1-one **5a** in the presence of a catalytic amount of both K_2CO_3 and Bu_3P at a higher temperature, which shows a novel example that the (*E*)-configuration of the carbon–carbon double bond could be transformed to the (*Z*)-configuration of the carbon–carbon double bond. It is worth mentioning here that compound **3a** could also be transformed to (*E*)-3-methyleneisoindolin-1-one **4a** and (*Z*)-3-methyleneisoindolin-1-one **5a** in the absence of catalyst, and compound **4a** could also be transformed to its isomer **5a**. There is an obvious intramolecular hydrogen bonding of the N–H group of **5a** in its crystal structure, and this could be the cause of the *Z*-stereoselectivity under higher temperature.¹³

With these results in hand, we first investigated the scope of alkynoates for the synthesis of (*Z*)-3-methyleneisoindolin-1-ones (Table 2). As shown in Table 2, all terminal alkynoates were proved to be applicable to this reaction and selectively gave product (*Z*)-3-methyleneisoindolin-1-ones in good to excellent yields under the optimized reaction conditions. The alkynoates bearing both little alkyl units, namely, ethyl and methyl, and a branched bulky alkyl group, such as *t*-butyl, could give the products in 95, 99, and 99% yields, respectively (entries 1–3, Table 2). The nature of the substituent on the benzene ring of the benzyl propiolate did impact the yields. For example, for substrates with a halogen Cl or methoxy group attached on the benzene ring, the yields of the corresponding products were obtained in the yields of 98% and 86%, respectively (entries 5 and 8, Table 2). Gratifyingly, furan-2-ylmethyl propiolate also reacted smoothly with **2a** to give the desired (*Z*)-3-methyleneisoindolin-1-one **5i** in an excellent

Table 2. Bu_3P -Catalyzed Synthesis of *Z*-3-Methyleneisoindolin-1-ones^a



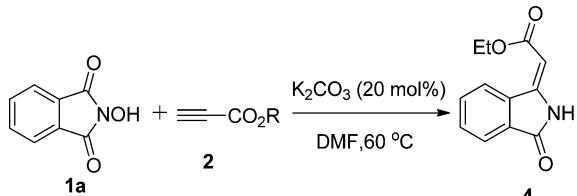
entry	R	time (h)	product	yield (%)
1	Et	2	5a	95
2	Me	2	5b	99
3	<i>t</i> -Bu	2	5c	99
4	PhCH ₂	2	5d	99
5	4-Cl-C ₆ H ₄ CH ₂	2	5e	98
6	4-Br-C ₆ H ₄ CH ₂	2	5f	94
7	4-Me-C ₆ H ₄ CH ₂	4	5g	90
8	4-MeO-C ₆ H ₄ CH ₂	6	5h	86
9	2-furyl-CH ₂	2	5i	99
10	Ph	6	5j	71

^aAll the reactions were carried out with **1a** (0.3 mmol) and **2** (0.36 mmol) at 150 °C in 1.0 mL of DMF under N_2 , and isolated yields were reported.

yield (entry 9, Table 2). It is worthy to note that phenyl propiolate could also proceed with **2a** to provide the corresponding (*Z*)-3-methyleneisoindolin-1-one **5j** in the yield of 71%. However, only the 3a-hydroxyisoxazolo[3,2-*a*]isoindol-8(3*aH*)-one products were formed, when β -substituted alkynoates were applied in the reaction.

We then investigated the scope of alkynoates for the synthesis of (*E*)-3-methyleneisoindolin-1-ones in the presence of K_2CO_3 (Table 3). As exemplified in Table 3, a wide array of alkynoates was suitable for the present strategy despite that some examples need to be careful about the reaction temperature. For example, the (*E*)-3-methyleneisoindolin-1-ones were effectively given at room temperature when methyl propiolate or 4-bromophenyl propiolate was used in this reaction; however, the (*Z*)-3-methyleneisoindolin-1-ones were formed as major products at 60 °C for 2 h (entries 2 and 6, Table 3). Notably, phenyl propiolate was also compatible with the standard reaction conditions but in a decreased yield (entry 10, Table 3).

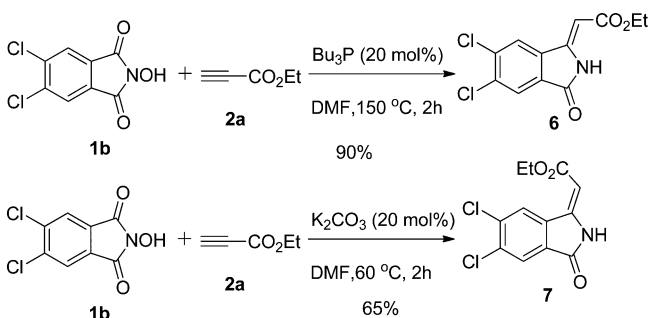
To examine the scope of 2-hydroxyisoindoline-1,3-dione, 5,6-dichloro-2-hydroxyisoindoline-1,3-dione (**1b**) was investigated. As expected, the (*Z*)-ethyl 2-(5,6-dichloro-3-oxoisoindolin-1-

Table 3. K_2CO_3 -Catalyzed Synthesis of *E*-3-Methyleneisindolin-1-ones^a


entry	R	time (h)	product	yield (%)
1	Et	4	4a	73
2 ^b	Me	12	4b	66
3	<i>t</i> -Bu	4	4c	62
4	PhCH ₂	4	4d	60
5	4-Cl-C ₆ H ₄ CH ₂	4	4e	62
6 ^b	4-Br-C ₆ H ₄ CH ₂	12	4f	60
7	4-Me-C ₆ H ₄ CH ₂	6	4g	60
8	4-MeO-C ₆ H ₄ CH ₂	6	4h	57
9	2-furyl-CH ₂	4	4i	63
10	Ph	6	4j	29

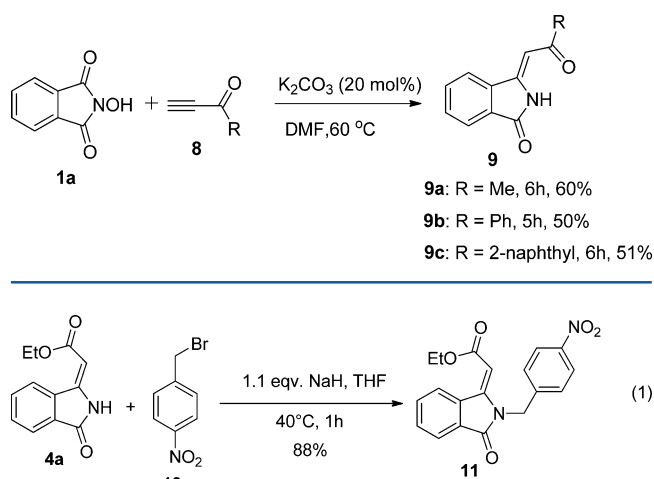
^aUnless noted otherwise, the reactions were carried out with **1a** (0.3 mmol) and **2** (0.36 mmol) at 60 °C in 1.0 mL of DMF under N₂, and isolated yields were reported. ^bRoom temperature.

ylidene)acetate (**6**) was formed in 90% yield when the reaction was performed in DMF at 150 °C for 2 h in the presence of Bu₃P, while the (*E*)-ethyl 2-(5,6-dichloro-3-oxoisindolin-1-ylidene)acetate (**7**) was formed in 65% yield when K₂CO₃ was used as catalyst in DMF at 60 °C for 2 h (Scheme 3).

Scheme 3. Reaction of 5,6-Dichloro-2-hydroxyisindoline-1,3-dione (**1b**) with Ethyl Propiolate (**2a**)

Subsequently, promoted by the successful base-catalyzed direct selective synthesis of 3-methyleneisindolin-1-ones with alkynoates, a range of alkyne ketones were also examined with respect to 2-hydroxyisindoline-1,3-dione **1a** to synthesize 3-methyleneisindolin-1-one compounds. To our surprise, the (*Z*)-3-methyleneisindolin-1-ones were given when alkyne ketones **8** and **1a** were disposed in DMF at 60 °C in the presence of K₂CO₃, while the reactions were very complicated when the reactions were carried out in the presence of Bu₃P under different temperatures. Both aromatic and aliphatic alkyne ketones could proceed smoothly to give (*Z*)-3-methyleneisindolin-1-ones in good yields (Scheme 4).

Next, the free N-H group of **4a** was attempted to protect with 1-(bromomethyl)-4-nitrobenzene (**10**) (2.0 equiv) in the presence NaH (2.0 equiv) in THF at 40 °C for 1 h. In the reaction, *N*-benzyl-protected compound **11** was obtained in 88% yield (eq 1).

Scheme 4. Reactions of 2-Hydroxyisindoline-1,3-dione (**1a**) with Alkyne Ketones (**8**)

In conclusion, we have developed an efficient and metal-free protocol to synthesize (*E*)-3-methyleneisindolin-1-ones and (*Z*)-3-methyleneisindolin-1-ones via base-catalyzed selective reactions of electron-deficient alkynes with *N*-hydroxyphthalimides. The present catalytic process provides a mild and general access to the (*E*)-3-methyleneisindolin-1-ones and (*Z*)-3-methyleneisindolin-1-ones, respectively. Especially, the structures of these products are attractive for potential drug discovery.

EXPERIMENTAL SECTION

General Methods. All reactions were performed in anhydrous solvents under a N₂ atmosphere. THF, Et₂O, and toluene were distilled from K and Na metal, respectively. DMF, CH₂Cl₂, and acetone were distilled from CaH₂. CH₃CN was distilled from P₂O₅. PE refers to petroleum ether (boiling range: 60–90 °C). Melting points were obtained on a melting point apparatus and are uncorrected. ¹H and ¹³C NMR spectra were recorded in DMSO-*d*₆ using a 300 MHz spectrometer; chemical shifts (δ) are given in parts per million, coupling constants (*J*) in Hz. High-resolution mass spectra were recorded in ESI mode on a QTOF MS spectrometer.

General Procedure for Reaction of Electron-Deficient Alkynes with *N*-Hydroxyphthalimide. To the solution of electron-deficient alkyne (0.36 mmol) and *N*-hydroxyphthalimide (0.3 mmol) in dry DMF (1 mL) was added Ph₃P (15.7 mg, 0.06 mmol). The resulting mixture was stirred at room temperature under a nitrogen atmosphere for the required period of time. After completion of the reaction as monitored by TLC, the reaction mixture was quenched with CH₂Cl₂ (15 mL), which was washed with water and brine successively, dried over MgSO₄, filtered, and concentrated in vacuo. Purification by flash chromatography (SiO₂; ethyl acetate/PE, 1:10–1:3) yielded the desired products.

(*E*)-Ethyl 2-(3-Oxoisindolin-1-ylidene)acetate (4a**).** 47.54 mg, 73% isolated yield; white solid. mp: 95–97 °C. IR (KBr) 3744, 3722, 3651, 2360, 2345, 1770, 1750, 1734, 1717, 1700, 1683, 1657, 1562, 1504, 1458, 794 cm⁻¹. ¹H NMR (300 MHz, DMSO-*d*₆): δ 10.34 (s, 1H), 8.11 (d, *J* = 7.3 Hz, 1H), 7.82 (d, *J* = 7.3 Hz, 1H), 7.72 (dt, *J* = 18.5, 7.2 Hz, 2H), 6.11 (s, 1H), 4.23 (q, *J* = 7.1 Hz, 2H), 1.28 (t, *J* = 7.1 Hz, 3H). ¹³C NMR (75 MHz, DMSO-*d*₆): δ 167.9, 166.2, 146.6, 136.5, 133.1, 131.7, 128.7, 123.1, 122.0, 92.0, 59.9, 14.1. ESIHRMS: Calcd for C₁₂H₁₂NO₃: (*M* + *H*)⁺ 218.0812. Found: *m/z* 218.0815.

(*E*)-Methyl 2-(3-Oxoisindolin-1-ylidene)acetate (4b**).** 40.20 mg, 66% isolated yield; white solid. mp: 197–199 °C; IR (KBr) 3167, 3113, 3086, 1744, 1715, 1643, 1694, 1632, 1607, 1454, 1373, 1302, 1273, 1157, 1115, 845, 773, 696 cm⁻¹. ¹H NMR (300 MHz, DMSO-*d*₆): δ 10.92 (s, 1H), 8.95 (d, *J* = 7.6 Hz, 1H), 7.83–7.66 (m, 3H), 5.80 (s, 1H), 3.72 (s, 3H). ¹³C NMR (75 MHz, DMSO-*d*₆): δ 169.3,

(KBr) 3327, 1744, 1682, 1667, 1422, 1273, 1294, 1273, 1196, 1179, 1125, 1063, 814, 768, 691 cm^{-1} . ^1H NMR (300 MHz, DMSO-*d*₆): δ 10.39 (s, 1H), 8.09 (d, *J* = 7.2 Hz, 1H), 7.80 (d, *J* = 7.2 Hz, 1H), 7.70 (dd, *J* = 10.7, 7.3 Hz, 2H), 7.31 (d, *J* = 7.8 Hz, 2H), 7.18 (d, *J* = 7.7 Hz, 2H), 6.13 (s, 1H), 5.19 (s, 2H), 2.29 (s, 3H). ^{13}C NMR (75 MHz, DMSO-*d*₆): δ 167.9, 166.0, 146.9, 137.2, 136.4, 133.1, 133.1, 131.7, 128.8, 128.7, 128.1, 123.1, 122.0, 91.7, 65.2, 20.6. ESIHRMS: Calcd for $\text{C}_{18}\text{H}_{16}\text{NO}_3$: (M + H)⁺ 294.1125. Found: *m/z* 294.1131.

(*Z*)-4-Methoxybenzyl 2-(3-Oxoisoindolin-1-ylidene)acetate (**5h**). 79.75 mg, 86% isolated yield; white solid. mp: 173–175 °C. IR (KBr) 3339, 1742, 1682, 1667, 1660, 1651, 1645, 1634, 1514, 1470, 1120, 1173, 1060, 822, 168, 694 cm^{-1} . ^1H NMR (300 MHz, DMSO-*d*₆): δ 10.39 (s, 1H), 8.09 (d, *J* = 7.5 Hz, 1H), 7.81 (d, *J* = 7.0 Hz, 1H), 7.70 (dd, *J* = 10.6, 7.4 Hz, 2H), 7.37 (d, *J* = 8.5 Hz, 2H), 6.94 (d, *J* = 8.5 Hz, 2H), 6.13 (s, 1H), 5.17 (s, 2H), 3.75 (s, 3H). ^{13}C NMR (75 MHz, DMSO-*d*₆): δ 168.5, 166.6, 159.7, 147.4, 137.0, 133.6, 132.3, 130.6, 129.2, 128.6, 123.7, 122.5, 114.3, 92.3, 65.8, 55.6. ESIHRMS: Calcd for $\text{C}_{18}\text{H}_{15}\text{NNaO}_4$: (M + Na)⁺ 332.0893. Found: *m/z* 332.0896.

(*Z*)-Furan-2-ylmethyl 2-(3-Oxoisoindolin-1-ylidene)acetate (**5i**). 79.92 mg, 99% isolated yield; white solid. mp: 119–121 °C. IR (KBr) 3426, 3333, 1746, 1732, 1688, 1665, 1645, 1422, 1198, 1177, 1126, 1062, 780, 687 cm^{-1} . ^1H NMR (300 MHz, DMSO-*d*₆): δ 10.98 (s, 1H), 8.95 (d, *J* = 7.6 Hz, 1H), 7.84–7.66 (m, 4H), 6.58 (d, *J* = 3.0 Hz, 1H), 6.48 (s, 1H), 5.77 (s, 1H), 5.19 (s, 2H). ^{13}C NMR (75 MHz, DMSO-*d*₆): δ 168.5, 166.3, 149.9, 147.8, 144.1, 136.9, 133.6, 132.3, 129.2, 123.7, 122.6, 111.3, 111.2, 91.8, 57.8. ESIHRMS: Calcd for $\text{C}_{15}\text{H}_{12}\text{NO}_4$: (M + H)⁺ 270.0761. Found: *m/z* 270.0764.

(*Z*)-Phenyl 2-(3-Oxoisoindolin-1-ylidene)acetate (**5j**). 56.46 mg, 71% isolated yield; white solid. mp: 134–136 °C. IR (KBr) 3242, 1713, 1694, 1643, 1634, 1275, 1202, 1153, 1117, 1096, 1057, 692 cm^{-1} . ^1H NMR (300 MHz, CDCl_3): δ 9.58 (s, 1H), 4.8 (d, *J* = 7.3 Hz, 1H), 7.78 (d, *J* = 4.4 Hz, 1H), 7.68 (dt, *J* = 12.1, 4.4 Hz, 2H), 7.44 (t, *J* = 4.6 Hz, 2H), 7.28 (dd, *J* = 5.4, 3.1 Hz, 1H), 7.19 (d, *J* = 4.9 Hz, 2H), 6.02 (s, 1H). ^{13}C NMR (75 MHz, CDCl_3): δ 167.9, 166.1, 150.4, 149.2, 136.3, 133.0, 131.9, 129.4, 126.0, 124.2, 121.5, 121.2, 114.9, 90.4. ESIHRMS: Calcd for $\text{C}_{16}\text{H}_{12}\text{NO}_3$: (M + H)⁺ 266.0812. Found: *m/z* 266.0816.

(*Z*)-Ethyl 2-(5,6-Dichloro-3-oxoisoindolin-1-ylidene)acetate (**6**). 76.95 mg, 90% isolated yield; yellow solid. mp: 194–196 °C. IR (KBr) 3390, 3381, 3242, 1737, 1732, 1694, 1682, 1651, 1645, 1634, 1614, 1557, 1537, 1505, 1271, 1067, 906, 854 cm^{-1} . ^1H NMR (300 MHz, DMSO-*d*₆): δ 10.53 (s, 1H), 8.49 (s, 1H), 8.01 (s, 1H), 6.20 (s, 1H), 4.21 (q, *J* = 6.9 Hz, 2H), 1.26 (t, *J* = 7.0 Hz, 3H). ^{13}C NMR (75 MHz, DMSO-*d*₆): δ 167.9, 167.7, 146.4, 138.3, 137.9, 136.3, 130.8, 127.0, 126.3, 95.9, 62.0, 16.0. ESIHRMS: Calcd for $\text{C}_{12}\text{H}_9\text{Cl}_2\text{N}_1\text{NaO}_3$: (M + Na)⁺ 307.9852. Found: *m/z* 307.9857.

(*E*)-Ethyl 2-(5,6-Dichloro-3-oxoisoindolin-1-ylidene)acetate (**7**). 55.58 mg, 65% isolated yield; white solid. mp: 215–217 °C. IR (KBr) 3200, 3109, 1743, 1732, 1713, 1694, 1643, 1634, 1607, 1393, 1280, 1145, 1105, 1036, 910, 850. ^1H NMR (300 MHz, DMSO-*d*₆): δ 11.18 (s, 1H), 9.10 (s, 1H), 7.95 (s, 1H), 5.77 (s, 1H), 4.18 (q, *J* = 7.0 Hz, 2H), 1.26 (t, *J* = 7.1 Hz, 3H). ^{13}C NMR (75 MHz, DMSO-*d*₆): δ 167.3, 167.2, 147.9, 137.5, 136.6, 135.0, 133.3, 131.2, 126.5, 101.5, 62.1, 15.9. ESIHRMS: Calcd for $\text{C}_{12}\text{H}_{10}\text{Cl}_2\text{N}_1\text{O}_3$: (M + H)⁺ 286.0032. Found: *m/z* 286.0037.

(*Z*)-3-(2-Oxopropylidene) Isoindolin-1-one (**9a**). 33.67 mg, 60% isolated yield; white solid. mp: 116–118 °C. IR (KBr) 3443, 3426, 3347, 3333, 1728, 1694, 1682, 1674, 1614, 1607, 1360, 1227, 1098, 760, 694 cm^{-1} . ^1H NMR (300 MHz, DMSO-*d*₆): δ 10.55 (s, 1H), 8.01 (d, *J* = 7.5 Hz, 1H), 7.80 (m, 2H), 7.75–7.68 (m, 1H), 6.52 (s, 1H), 2.28 (s, 3H). ^{13}C NMR (75 MHz, DMSO-*d*₆): δ 200.2, 170.4, 146.4, 138.8, 136.1, 135.0, 133.8, 125.3, 124.7, 123.7, 101.4, 32.8. ESIHRMS: Calcd for $\text{C}_{11}\text{H}_{10}\text{NO}_2$: (M + H)⁺ 188.0707. Found: *m/z* 188.0706.

(*Z*)-3-(2-Oxo-2-phenylethylidene) Isoindolin-1-one (**9b**). 37.36 mg, 50% isolated yield; yellow solid. mp: 165–167 °C. IR (KBr) 3333, 1720, 1651, 1599, 1300, 1227, 1018, 760, 710 cm^{-1} . ^1H NMR (300 MHz, DMSO-*d*₆): δ 10.89 (s, 1H), 8.34 (d, *J* = 7.5 Hz, 1H), 8.19 (d, *J* = 7.4 Hz, 2H), 7.81 (dd, *J* = 14.9, 7.4 Hz, 2H), 7.75–7.62 (m, 2H), 7.58 (t, *J* = 7.3 Hz, 2H), 7.39 (s, 1H). ^{13}C NMR (75 MHz, DMSO-*d*₆): δ 191.7, 170.6, 149.4, 139.7, 139.0, 135.1, 134.9, 133.9,

130.6, 130.3, 130.0, 125.2, 124.4, 97.5. ESIHRMS: Calcd for $\text{C}_{16}\text{H}_{12}\text{NO}_2$: (M + H)⁺ 250.0867. Found: *m/z* 250.0863.

(*Z*)-3-(2-(Naphthalen-2-yl)-2-oxoethylidene) Isoindolin-1-one (**9c**). 45.76 mg, 51% isolated yield; white solid. mp: 211–213 °C. IR (KBr) 3426, 3412, 3381, 3352, 1715, 1694, 1645, 1593, 810, 768 cm^{-1} . ^1H NMR (300 MHz, DMSO-*d*₆): δ 10.95 (s, 1H), 8.96 (s, 1H), 8.41 (d, *J* = 7.5 Hz, 1H), 8.18 (t, *J* = 8.6 Hz, 2H), 8.06 (dd, *J* = 15.2, 8.0 Hz, 2H), 7.86 (t, *J* = 8.4 Hz, 2H), 7.81–7.63 (m, 3H), 7.58 (s, 1H). ^{13}C NMR (75 MHz, DMSO-*d*₆): δ 191.5, 170.7, 149.3, 139.1, 137.1, 136.9, 135.1, 134.1, 133.9, 131.8, 131.4, 130.5, 130.4, 130.3, 129.6, 128.8, 125.7, 125.3, 124.4, 97.7. ESIHRMS: Calcd for $\text{C}_{20}\text{H}_{14}\text{NO}_2$: (M + H)⁺ 300.1019. Found: *m/z* 300.1026.

(*E*)-Ethyl 2-(2-(4-Nitrobenzyl)-3-oxoisoindolin-1-ylidene)acetate (**11**). 98.76 mg, 88% isolated yield; white solid. mp: 138–140 °C. IR (KBr) 3389, 3381, 1720, 1709, 1694, 1634, 1610, 1520, 1342, 1323, 1188, 1155, 1101, 829, 772, 708 cm^{-1} . ^1H NMR (300 MHz, DMSO-*d*₆): δ 8.94 (d, *J* = 7.4 Hz, 1H), 8.20 (d, *J* = 8.4 Hz, 2H), 7.92 (d, *J* = 7.2 Hz, 1H), 7.86–7.66 (m, 2H), 7.51 (d, *J* = 8.4 Hz, 2H), 5.78 (s, 1H), 5.22 (s, 2H), 4.17 (dd, *J* = 13.8, 6.9 Hz, 2H), 1.22 (t, *J* = 6.9 Hz, 3H). ^{13}C NMR (75 MHz, DMSO-*d*₆): δ 168.3, 167.0, 148.6, 145.9, 136.4, 135.5, 134.9, 133.7, 129.6, 129.3, 125.7, 125.1, 124.7, 101.3, 62.1, 43.6, 15.9. ESIHRMS: Calcd for $\text{C}_{19}\text{H}_{16}\text{N}_2\text{NaO}_5$: (M + H)⁺ 375.0951. Found: *m/z* 375.0955.

■ ASSOCIATED CONTENT

Supporting Information

The ORTEP diagrams of **4a** and **5a**, ^1H and ^{13}C NMR spectra of the products as well as the X-ray crystallographic data (CIF files) of **4a** and **5a**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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■ REFERENCES

- (1) Blasko, G.; Gula, D. J.; Shamma, M. *J. Nat. Prod.* **1982**, *45*, 105–122.
- (2) Valencia, E.; Fajardo, V.; Freyer, A. J.; Shamma, M. *Tetrahedron Lett.* **1985**, *26*, 993–996.
- (3) (a) Kato, Y.; Takemoto, M.; Achiwa, K. *Chem. Pharm. Bull.* **1993**, *41*, 2003–2006. (b) Kato, Y.; Ebike, H.; Achiwa, K.; Ashizawa, N.; Kurihara, T.; Kobayashi, F. *Chem. Pharm. Bull.* **1990**, *38*, 2060–2062.
- (4) Laboratori Baldacci, S. P. A. (Italy). Jpn. Patent 59046268, 1984.
- (5) (a) Lu, W.-D.; Lin, C.-F.; Wang, C.-J.; Wang, S.-J.; Wu, M.-J. *Tetrahedron Lett.* **2002**, *58*, 7315–7319. (b) Koradin, C.; Dohle, W.; Rodriguez, A. L.; Schmid, B.; Knochel, P. *Tetrahedron* **2003**, *59*, 1571–1587. (c) Kundu, N. G.; Khan, M. W. *Tetrahedron Lett.* **1997**, *38*, 6937–6940. (d) Yao, T.; Larock, R. C. *J. Org. Chem.* **2005**, *70*, 1432–1437. (e) Wang, C.; Sun, C.; Weng, F.; Gao, M.; Liu, B.; Xu, B. *Tetrahedron Lett.* **2011**, *52*, 2984–2989.
- (6) Flitsch, W.; Peters, H. *Tetrahedron Lett.* **1969**, 1161–1162.
- (7) (a) Pigeon, P.; Decroix, B. *Tetrahedron Lett.* **1996**, *37*, 7707–7710. (b) Daich, A.; Marchalin, S.; Pigeon, P.; Decroix, B. *Tetrahedron Lett.* **1998**, *39*, 9187–9190. (c) Chihab-Eddine, A.; Daich, A.; Jilale, A.; Decroix, B. *Heterocycles* **2002**, *58*, 449–456. (d) Kato, Y.; Ebike, H.; Achiwa, K.; Ashizawa, N.; Kurihara, T.; Kobayashi, F. *Chem. Pharm. Bull.* **1990**, *38*, 2060–2062.

(8) (a) Couture, A.; Deniau, E.; Grandclaoudon, P. *Tetrahedron* **1997**, *53*, 10313–10330. (b) Couture, A.; Deniau, E.; Grandclaoudon, P.; Hoarau, C. *Tetrahedron* **2000**, *56*, 1491–1499. (c) Couture, A.; Deniau, E.; Grandclaoudon, P.; Hoarau, C.; Rys, V. *Tetrahedron Lett.* **2002**, *43*, 2207–2210. (d) Couture, A.; Deniau, E.; Grandclaoudon, P.; Rybalko-Rosen, H.; Léonce, S.; Pfeiffer, B.; Renard, P. *Bioorg. Med. Chem. Lett.* **2002**, *12*, 3557–3559. (e) Rys, V.; Couture, A.; Deniau, E.; Grandclaoudon, P. *Tetrahedron* **2003**, *59*, 6615–6619. (f) Moreau, A.; Couture, A.; Deniau, E.; Grandclaoudon, P. *J. Org. Chem.* **2004**, *69*, 4527–4530.

(9) Hendi, M. S.; Natalie, K. J., Jr.; Hendi, S. B.; Campbell, J. A.; Greenwood, T. D.; Wolfe, J. F. *Tetrahedron Lett.* **1989**, *30*, 275–278.

(10) (a) Kundu, N. G.; Khan, M. W. *Tetrahedron Lett.* **1997**, *38*, 6937–6940. (b) Kundu, N. G.; Khan, M. W.; Mukhopadhyay, R. *Tetrahedron* **1999**, *55*, 12361–12376. (c) Yao, T.; Larock, R. C. *J. Org. Chem.* **2005**, *70*, 1432–1437.

(11) (a) Khan, M. W.; Kundu, N. G. *Synlett* **1997**, 1435–1437.

(b) Sashida, H.; Kawamukai, A. *Synthesis* **1999**, 1145–1148.

(c) Kundu, N. G.; Khan, M. W. *Tetrahedron* **2000**, *56*, 4777–4792.

(d) Koradin, C.; Dohle, W.; Rodriguez, A. L.; Schmid, B.; Knochel, P. *Tetrahedron* **2003**, *59*, 1571–1587.

(e) Uozumi, Y.; Kawasaki, N.; Mori, E.; Mori, M.; Shibasaki, M. *J. Am. Chem. Soc.* **1989**, *111*, 3725–

3727. (f) Garcia, A.; Rodriguez, D.; Castedo, L.; Saa, C.; Dominguez, D. *Tetrahedron Lett.* **2001**, *42*, 1903–1905. (g) Comins, D. L.; Joseph, S. P.; Zhang, Y.-M. *Tetrahedron Lett.* **1996**, *37*, 793–796. (h) Wrigglesworth, J. W.; Cox, B.; Lloyd-Jones, G. C.; Booker-Milburn, K. I. *Org. Lett.* **2011**, *13*, 5326–5329. (i) Mancuso, R.; Ziccarelli, I.; Armentano, D.; Marino, N.; Giofre, S. V.; Gabriele, B. *J. Org. Chem.* **2014**, *79*, 3506–3518. (j) Liu, X.; Hii, K. K. *Eur. J. Org. Chem.* **2010**, 5181–5189. (k) Zhou, B.; Hou, W.; Yang, Y.; Li, Y. *Chem.—Eur. J.* **2013**, *19*, 4701–4706. (l) Patureau, F. W.; Besset, T.; Frohlich, R.; Glorius, F. C. *R. Chim.* **2012**, *15*, 1081–1085.

(12) Zhou, Q.-F.; Chu, X.-P.; Ge, F.-F.; Wang, Y.; Lu, T. *Adv. Synth. Catal.* **2013**, *355*, 2787–2792.

(13) CCDC 1032756 and CCDC 1032759 contain the supplementary crystallographic data for compounds **4a** and **5a**, respectively.

These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.